Commensurate Columnar Composite Compounds: Synthesis and Structure of Ba₁₅Zr₁₄Se₄₂ and Sr₂₁Ti₁₉Se₅₇

Louis J. Tranchitella,^{1a} James C. Fettinger,^{1a} Peter K. Dorhout,^{1b} Pamela M. Van Calcar,^{1b} and Bryan W. Eichhorn*,1a

> Department of Chemistry and Biochemistry University of Maryland, College Park, Maryland 20742 Department of Chemistry, Colorado State University Fort Collins, Colorado 80523 Received July 21, 1997 Revised Manuscript Received June 11, 1998

Among the ternary sulfide and selenide AMQ₃ compounds (A = alkaline earth ion, rare earth ion, M = early transition metal, Q = S, Se), the hexagonal BaNiO₃-type²⁻⁵ and the LaCrS₃-type "misfit layer" structures⁶⁻⁸ are most common.⁹ Compounds reported to have the pseudo-one-dimensional BaNiO3-type structure include $ATiS_3$ (A = Ba, Sr), BaZrSe₃, and BaVQ₃ (Q = S, Se).²⁻⁴ The structures are characterized by ${}^{1}_{\infty}$ [MQ_{6/2}] chains of face-sharing octahedra (see 1) extending along the c-axis of the hexagonal cell (prototypical crystal symmetry P63/mmc). The



A-site ions are 12-coordinate (see 2) and fill the interchain channels as shown by the projection down the c-axis given in 3. In this structure type, the number of A and M atoms are equivalent and, therefore, the periodicities of the ${}^{1}_{m}[A]^{2+}$ and ${}^{1}_{m}[MQ_{6/2}]^{2-}$ chains are matched. An interesting new distorted version of this structure was recently found in the $CsTaQ_3$ phases (Q = S, Se, Te) in which the Ta^{5+} ions are displaced from the centers of the Q₆ octahedra.¹⁰

The elegant studies of Saeki, Onoda, and co-workers previously showed that some of these BaNiO₃-type compounds are slightly nonstoichiometric with actual compositions of $A_{1+x}TiS_3$ (where A = Ba, Sr; x < 0.19).^{11–13} The compounds were shown to be highly adaptive with an infinite number of phases within given ranges of $x^{.11,12}$ The only structural data for these compounds, Rietveld analyses of powder X-ray data for Sr_{1.145}TiS₃ and Sr_{1.19}-TiS₃, revealed a new type of columnar composite structure in

- (2) Aslanov, L. A.; Kovba, M. Russ. J. Inorg. Chem. 1964, 9, 1317.
 (3) Aslanov, L. A. Russ. J. Inorg. Chem. 1964, 9, 1090.

- (4) Hahn, H.; Mutschke, U. Z. Anorg. Allg. Chem. 1956, 288, 269.
 (5) Matsuura, K.; Nakamizo, T.; Wada, T.; Yamauchi, H.; Tanaka, S. Jpn. J. Appl. Phys. 1993, 32, 662.
- (6) Incomensurate Sandwiched Layered Compounds; Meerschaut, A., Ed.;
 Trans. Technol. Publishers: Zurich, 1992; Vol. 100–101.
 (7) Makovicky, E.; Hyde, B. G. Mater. Sci. Forum 1992, 100/101, 1.
- (a) Wiegers, G. A. Jpn. J. Appl. Phys. 1993, 32, 705.
 (b) Wiegers, G. A. Jpn. J. Appl. Phys. 1993, 32, 705.
 (c) Eichhorn, B. W. In Progress in Inorganic Chemistry; K. D. Karlin, Ed.; John Wiley & Sons: New York, 1994; Vol. 42, p 139.
 (10) Pell, M. A.; Vajenine, G. V. M.; Ibers, J. A. J. Am. Chem. Soc. 1997, 110.
- 119, 5186.
 - (11) Saeki, M.; Onoda, M. J. Solid State Chem. 1993, 102, 100. (12) Saeki, M.; Onoda, M. J. Solid State Chem. 1994, 112, 65.
- (13) Onoda, M.; Saeki, M.; Yamamoto, A.; Kato, K. Acta Crystallogr. 1993, B49, 929.

which the $\frac{1}{m}[\text{TiS}_{6/2}]^{2.29-}$ and $\frac{1}{m}[\text{Sr}]^{2+}$ chains are mutually incommensurate along the *c*-axis of a BaNiO₃-related cell.^{13,14} The incommensurate nature of the structure causes significant variations from the ideal BaNiO3-type hcp lattice. We recently isolated single crystals of two members in this class of compounds; namely, "BaZrSe3" and "SrTiSe3". Herein we report the synthesis, structures, and preliminary characterization of these unusual chain compounds that contain commensurate rhombohedral crystal structures with large *c*-axes of 91.5 and 117.9 Å, respectively. The phases have actual compositions of Ba₁₅Zr₁₄Se₄₂ (or Ba_{1.07}-ZrSe₃) and Sr₂₁Ti₁₉Se₅₇ (or Sr_{1.11}TiSe₃) with mixtures of MSe_{6/2} octahedra (O) and distorted trigonal prisms (TP) in BaNiO₃-related structures.

Single phase Ba₁₅Zr₁₄Se₄₂ was prepared from BaSe, Zr, and Se in a 1:1:2 molar ratio at 1050 °C (80 h) with 40 wt % BaCl₂ flux. After the reaction, the flux was dissolved in water leaving thick black crystalline needles and a black powder in $\sim 1:1$ mass ratio. Sr₂₁Ti₁₉Se₅₇ was prepared in a similar manner from SrSe, Ti, and Se in a 1:1:2 molar ratio at 1050 °C (80 h) with 10 wt % SrCl₂ flux. Removal of the flux with water left small black needles (~30% of sample) of Sr₂₁Ti₁₉Se₅₇ along with a black powder containing unidentified impurities. The single crystals of both compounds have been characterized by EDX and singlecrystal X-ray diffraction (conventional 4-circle and CCD diffractometers) and preliminary EPR spectroscopic studies, magnetic susceptibility, and 4-probe resistivity measurements have been made on Ba₁₅Zr₁₄Se₄₂. Wavelength dispersive X-ray analysis (WDS) on several crystals of Ba₁₅Zr₁₄Se₄₂ (or Ba_{1.07}ZrSe₃) gave a consistent composition Ba_{1.12±0.02}Zr_{1.00±0.03}Se_{3.01±0.03} which is in excellent agreement with the refined X-ray compositions (see below). Attempts to prepare "stoichiometric" BaZrSe₃ or SrTiSe₃ were unsuccessful with or without added flux.

The structure of Ba15Zr14Se42 comprises chains of ${}^1_{\sim}[ZrSe_{6/2}]^{2.14-}$ and ${}^1_{\infty}[Ba]^{2+}$ in a $BaNiO_3\text{-related cell.}$ Unlike the parent BaNiO₃ structure, the repeat units for the chains are mismatched at 45.75 and 30.5 Å, respectively. In contrast to the $A_{1+x}TiS_3$ incommensurate columnar composite compounds, ^{11–13} the chains in $Ba_{15}Zr_{14}Se_{42}$ share a common repeat unit of 91.51 Å giving rise to a commensurate rhombohedral crystal structure, space group $R\bar{3}c$, with a = 12.4366(1) Å and c = 91.5069(2)Å.¹⁵ The high residuals presumably reflect the abnormally long *c*-axis, the weakness of the diffraction data and the fact that the structure may be on the verge of becoming incommensurate. Several frame photographs collected with a CCD detector confirmed the space group choice and did not reveal any spurious satellite reflections that would indicate an incommensurate lattice. Despite the high residuals, the structure is well behaved and the final metric parameters are quite reasonable.

The most interesting feature is the ${}^{1}_{\infty}$ [ZrSe_{6/2}]^{2.14-} chains with eight crystallographically distinct Zr atoms. The 45.75 Å $[Zr_{14}Se_{42}]^{30-}$ repeat unit of the chain contains 12 ZrSe_{6/2} octahedra (O) and two distorted trigonal prisms (TP) that order according to the following sequence: -O₁₁-TP-O-TP-O₁₁-TP-O-TP-O₁₁- (see Figure 1a). The Zr-Se contacts in both the O and TP polyhedra range from 2.612(4) to 2.769(4) Å with no statistical differences between the two types of polyhedra. The intrachain Zr-Zr separations range from 3.050(5) Å to 3.328(6) Å between adjacent octahedra whereas the O-TP separations are somewhat longer

^{(1) (}a) University of Maryland. (b) Colorado State University

⁽¹⁴⁾ Onoda, M.; Saeki, M. Jpn. J. Appl. Phys. **1993**, 32, 752. (15) Crystal Data. For Ba₁₅Zr₁₄Se₄₂: rhombohedral, R3c; a = 12.4366(1)and c = 91.5069(6) Å; V = 12257.1(2) Å³, Z = 6, $\rho_{calcd} = 5.408$ g/cm³; $\mu_{M_{6}Ka} = 27.51$ mm⁻¹; T = 298 K; $R_{1}(F) = 9.89$ for those 1564 reflections with $F_0 > 4\sigma F_0$, $R_1(F) = 11.58$ for all data and wR(F^2) = 28.25. For Sr₂₁-Ti₁₉Se₅₇: rhombohedral, $R\bar{3} a = 11.9517(5)$ and c = 117.90 (4) Å; V = 14584(5) Å³, Z = 6, $\rho_{calcd} = 4.953$ g/cm³; $\mu_{Mo K\alpha} = 34.22$ mm⁻¹. $R_i(F) = 8.34$ for those 1856 reflections with $F_0 > 4\sigma F_0$, $R_1(F) = 19.59$ for all data, and wR- $(F^2) = 29.10$. See the Supporting Information for full crystallographic details.



Figure 1. Graphical representations of the (**a**) $-O_{11}$ -TP-O-TP- O_{11} -TP-O-TP- chains in Ba₁₅Zr₁₄Se₄₂ and (**b**) $(-O_7$ -TP-O-TP)₃-O₅-TP-O-TP-(-O₇-TP-O-TP)₃-O₅-TP-O-TP-(-O₇-TP-O-TP)₃-O₅-TP-O-TP- chains in Sr₂₁Ti₁₉Se₅₇. The large filled circles represent the "TP" in the -TP-O-TP- units. (**c**) An approximate (001) view down the "O₁₁" portion of the chain defined by Zr(6)-Zr(3)-...-Zr(3)-Zr(6). (**d**) An approximate (001) view down the "O-TP-O-TP-O" portion of the chain defined by Zr(6)-Zr(4)-Zr(2)-Zr(4)-Zr(6). (**e**) A view down the C₃-axis of a TiSe₆ distorted trigonal prism showing the nonideal dihedral angle (α). (**f**) An ORTEP drawing of an TP-O-TP (Ti₃Se₁₂)¹²⁻ unit in Sr₂₁Ti₁₉Se₅₇ with only the Ti atoms labeled.

at 3.386(6) Å and 3.413(5) Å. The 11 consecutive face-sharing ZrSe_{6/2} octahedra (the O₁₁ segments) are highly regular as evidenced by the view down the chain axis in Figure 1c. The eclipsed Se atoms show a periodic modulation in the O₁₁ region which is similar to that observed in Sr_{1.145}TiS₃.^{13,14} A similar view down the O-TP-O-TP-O segment (Figure 1d) shows the highly irregular arrangement of ZrSe_{6/2} polyhedra in this region. The TP geometry is in fact intermediate between an octahedron and a trigonal prism as evidenced by the 36.02° (av) Zr(4)–Se-(3)–Se(2)–Zr(4) dihedral angle (α in Figure 1e). This value is intermediate to those of an ideal octahedron (60°) and a trigonal prism (0°).

The average Zr oxidation state in the $[Zr_{14}Se_{42}]^{30-}$ segment is +3.86; however, if the Zr ions are partitioned into integral valencies, one finds 12 Zr^{IV} and 2 Zr^{III} ions per segment. One interpretation is that the two trigonal prismatic ZrSe_{6/2} units in each segment contain the valence localized Zr^{III} ions. This proposal is supported by EPR and resistivity measurements (see below) but is not obvious from the Zr–Se distances.

Sr₂₁Ti₁₉Se₅₇ also possesses a commensurate rhombohedral structure, space group $R\overline{3}$, with a = 11.9517(5) Å and c = 117.90-(4) Å.¹⁵ The compound contains a 117.9 Å [Ti₃₈Se₁₁₄]^{84–} repeat unit in the $\frac{1}{\infty}$ [TiSe₆₂]^{2.21–} chains (Figure 1b) with TP–O–TP units (Figure 1f) virtually identical to those in Ba₁₅Zr₁₄Se₄₂. However, the number of octahedra separating the TP–O–TP units varies from five to seven according to the following sequence: (-O₇-TP-O-TP)₃-O₅-TP-O-TP-(-O₇-TP-O-TP)₃-O₅-TP-O-TP-

The average Ti oxidation state in the $[Ti_{38}Se_{114}]^{84-}$ repeat unit is +3.79 which gives 8 Ti^{III} and 30 Ti^{IV} when partitioned into integral valencies. Once again, the number of distorted trigonal prismatic and octahedral sites in the chain (8 TP, 30 O) mirrors the number of M^{III} and M^{IV} centers, respectively, suggesting that the d¹ Ti³⁺ ions are localized at the TP sites. As with the Ba₁₅Zr₁₄-Se₄₂ phase, this proposal is not obvious from the structural parameters. The Ti–Se distances in the TP units range from 2.536 to 2.642 Å (2.583 Å, av) whereas the Ti–Se distances in the octahedra range from 2.496 to 2.652 Å (2.567 Å, av). The Ti– Ti separations between octahedra range from 2.935 to 3.064 Å with longer separations between O–TP units (3.1110–3.295 Å). The dihedral angles, α , for the trigonal prisms span 19.2 to 30.37°.

Ba₁₅Zr₁₄Se₄₂ is EPR active showing a broad signal at -170 °C with $g_{\parallel} = 2.056$ and $g_{\perp} = 1.998$. Magnetic susceptibility experiments show Curie–Weiss behavior above 150 K with $\mu_{\rm eff} = 1.63 \ \mu_{\rm B}$ per TP unit assuming all other ions are diamagnetic. Four-probe resistivity measurements (single crystal) between -180 and 25 °C show intrinsic semiconducting behavior with

 $E_{\rm g} = 0.22$ eV. This measured gap is in good agreement with that reported for "BaZrSe₃" by Aslonov³ suggesting that the compounds are the same. These data are consistent with the interpretation suggested by the stoichiometry and structure; namely, that the Zr^{III} centers are localized on the distorted trigonal prismatic sites of the [Zr₁₄Se₄₂]³⁰⁻ segments. We have been unable to prepare single-phase Sr₂₁Ti₁₉Se₅₇ and have therefore not investigated its properties. It is interesting to note that titanium oxide perovskites of similar stoichiometry and oxidation state are known to be metallic (i.e., metallic Nd_{0.20}Ba_{0.80}TiO₃, av Ti oxidation state +3.8)^{16,17} as are the one-dimensional BaNiO₃-type sulfides BaNbS₃ (>620 K)⁵ and BaVS₃.¹⁸

The origin of the alkaline earth "excess" in the $A_{1+x}MSe_3$ phases and the distortions in the $\frac{1}{10}[MSe_{6/2}]^{(2+n)-}$ chains appear to stem from the reluctance of the A^{2+} ions (A = Ba, Sr) to reside in the 12-coordinate hcp lattice site required by an undistorted BaNiO₃ structure type. As with the BaNiO₃ structure type, there are three $\frac{1}{0}[MSe_{6/2}]^{(2+n)-}$ chains defining the channels in each structure that host the A^{2+} ions (see **3**). The TP–O–TP "kinks" in the respective $\frac{1}{0}[MSe_{6/2}]^{(2+n)-}$ chains effectively lower the A^{2+} coordination number at that point in the channel. By offsetting the chains relative to each other along the A^{2+} channels, all of the A^{2+} ions achieve lower coordination numbers but are in high concentration relative to the parent BaNiO₃ type. In Ba₁₅-Zr₁₄Se₄₂, there are three crystallographically distinct pseudo-10-coordinate Ba²⁺ ions with Ba–Se contacts ranging from 3.201(6) to 4.138(3) Å. In Sr₂₁Ti₁₉Se₅₇, there is one 8-coordinate and six 9-coordinate Sr²⁺ ions with Sr–Se contacts ranging from 2.71-(1) to 3.70(1) Å. Further studies are in progress.

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Supporting Information Available: X-ray crystallographic tables and details (115 pages, print/PDF). See any current masthead page for ordering information and Web access instructions.

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⁽¹⁶⁾ Eylem, C.; Sághi-Szabó, G.; Chen, B.-H.; Eichhorn, B.; Peng, J.-L.; Greene, R.; Salamanca-Riba, L.; Nahm, S. Chem. Mater. 1992, 4, 1038. (17) Eylem, C.; Eichhorn, B. W.; Ju, H. L.; Greene, R. J. Solid State Chem. DOC 1010 (2010) (20

^{1995, 114, 164.} (18) Matsuura, K.; Wada, T.; Nakamizo, T.; Yamauchi, H.; Tanaka, S. Phys. Rev. 1991, B43, 13118.